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#### Key indicators

Single-crystal X-ray study

$T = 291$  K

Mean  $\sigma(\text{C}-\text{C}) = 0.004$  Å

$R$  factor = 0.023

$wR$  factor = 0.048

Data-to-parameter ratio = 16.1

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

## catena-Poly[[aqua[3-(3-pyridyl)acrylato]europium(III)]-di- $\mu$ -3-(3-pyridyl)acrylato]

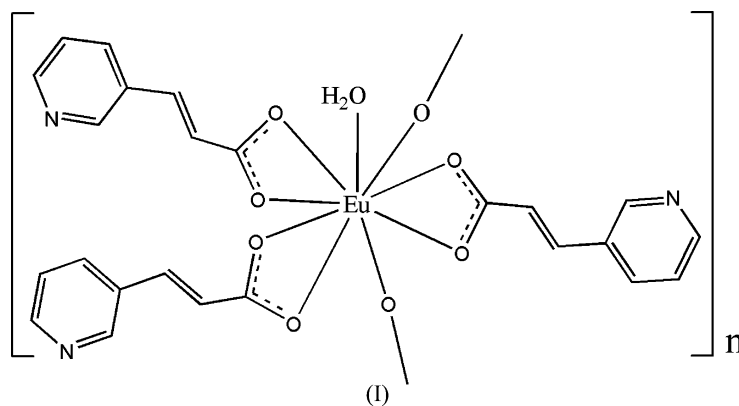
In the title compound,  $[\text{Eu}(\text{C}_8\text{H}_6\text{NO}_2)_3(\text{H}_2\text{O})]_n$ , the europium(III) ion is coordinated by eight carboxylate O atoms and one water molecule. The carboxylate ligands bridge pairs of europium(III) ions, forming a zigzag chain along [100]; hydrogen bonds link the chains into sheets parallel to (001).

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#### Comment

In recent years, research on metal–organic coordination polymers has been rapidly expanding because of their structural diversity and their potential applications as functional materials (Moulton *et al.*, 2001; Li *et al.*, 1998). The bifunctional ligand 3-pyridylacrylic acid (HTPA) is a potential multidentate ligand, and several types of complexes of HTPA have been studied (Zhang *et al.*, 2000; Ayyappan *et al.*, 2001; Gunning *et al.*, 2005). Until now, however, only a few crystallographic studies of 4f-block metal complexes of HTPA have been reported (Gunning *et al.*, 2005).



Here, we report the synthesis and structure of the title complex,  $[\text{Eu}(\text{TPA})_3(\text{H}_2\text{O})]_n$  (I) (Fig. 1), whose structure consists of a repeating unit of formula  $[\text{Eu}(\text{C}_8\text{H}_6\text{NO}_2)_3(\text{H}_2\text{O})]$ . Each  $\text{Eu}^{\text{III}}$  centre is coordinated by eight carboxylate O atoms and one water molecule. The europium(III) ions are joined into a coordination polymer chain along [100], reinforced by  $\text{O}-\text{H}\cdots\text{O}$  hydrogen bonds (Table 1). Adjacent [100] chains are linked by  $\text{O}-\text{H}\cdots\text{N}$  hydrogen bonds, forming sheets parallel to (001) (Fig. 2).

#### Experimental

A mixture of  $\text{EuCl}_3\cdot 6\text{H}_2\text{O}$  (0.2 mmol), 3-pyridylacrylic acid (0.2 mmol),  $\text{H}_2\text{O}$  (5 ml), and 0.65 M NaOH aqueous solution (0.2 ml, 0.13 mmol) was sealed in a 25 ml Teflon-lined stainless reactor and heated at 393 K for 72 h under autogenous pressure, then cooled at a rate of 10 K every 3 h to 373 K, followed by slow cooling to room

temperature, when a few colourless crystals were obtained. Analysis: found C 47.2, H 3.3, N 7.2%;  $C_{24}H_{20}EuN_3O_7$  requires C 46.9, H 3.3, N 6.8%.

Crystal data

[Eu(C<sub>8</sub>H<sub>6</sub>NO<sub>2</sub>)<sub>3</sub>(H<sub>2</sub>O)]  
*M<sub>r</sub>* = 614.39  
 Triclinic, *P* $\bar{1}$   
*a* = 6.2459 (6) Å  
*b* = 12.7698 (13) Å  
*c* = 15.7220 (16) Å  
 $\alpha$  = 111.7710 (10)°  
 $\beta$  = 90.3200 (10)°  
 $\gamma$  = 95.1760 (10)°  
*V* = 1158.7 (2) Å<sup>3</sup>  
*Z* = 2  
*D<sub>x</sub>* = 1.761 Mg m<sup>-3</sup>  
 Mo *K*α radiation  
 $\mu$  = 2.76 mm<sup>-1</sup>  
*T* = 291 (2) K  
 Block, colourless  
 0.15 × 0.09 × 0.08 mm

Data collection

Bruker SMART CCD area-detector diffractometer  
 $\varphi$  and  $\omega$  scans  
 Absorption correction: multi-scan (SADABS; Sheldrick, 1996)  
*T<sub>min</sub>* = 0.675, *T<sub>max</sub>* = 0.799  
 10024 measured reflections  
 5207 independent reflections  
 4755 reflections with *I* > 2σ(*I*)  
*R<sub>int</sub>* = 0.021  
 $\theta_{max}$  = 27.5°

Refinement

Refinement on *F*<sup>2</sup>  
*R*[*F*<sup>2</sup> > 2σ(*F*<sup>2</sup>)] = 0.023  
*wR*(*F*<sup>2</sup>) = 0.048  
*S* = 1.04  
 5207 reflections  
 324 parameters  
 H-atom parameters constrained  
 $w = 1/[\sigma^2(F_o^2) + (0.0184P)^2 + 0.2729P]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{max} = 0.002$   
 $\Delta\rho_{max} = 0.43 \text{ e \AA}^{-3}$   
 $\Delta\rho_{min} = -0.50 \text{ e \AA}^{-3}$

Table 1

Hydrogen-bond geometry (Å, °).

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
O7—H1W...O3 <sup>i</sup>	0.821 (17)	1.946 (17)	2.765 (3)	175 (3)
O7—H2W...N1 <sup>ii</sup>	0.840 (17)	1.901 (18)	2.731 (3)	170 (4)

Symmetry codes: (i) *x* - 1, *y*, *z*; (ii) -*x* + 1, -*y* + 2, -*z* + 1.

All H atoms were positioned geometrically and treated as riding atoms with C—H = 0.93 Å and *U<sub>iso</sub>*(H) = 1.2*U<sub>eq</sub>*(C).

Data collection: SMART (Bruker, 1998); cell refinement: SAINT (Bruker, 1998); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Bruker, 1998); software used to prepare material for publication: SHELXTL.

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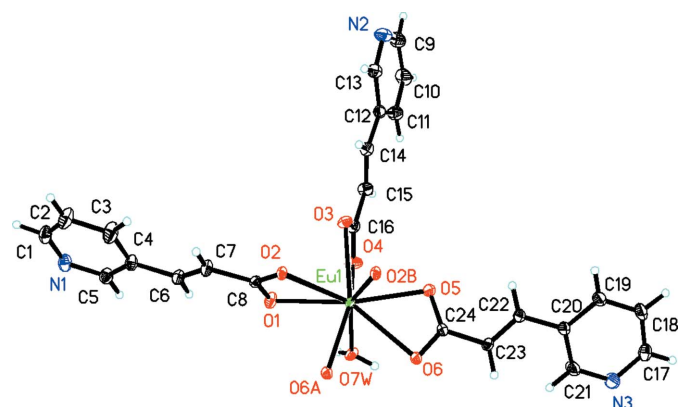


Figure 1

The coordination environment of the Eu<sup>III</sup> ion in (I), with displacement ellipsoids drawn at the 30% probability level. [Symmetry codes: (A) -*x*, 1 - *y*, 1 - *z*; (B) 1 - *x*, 1 - *y*, 1 - *z*.]

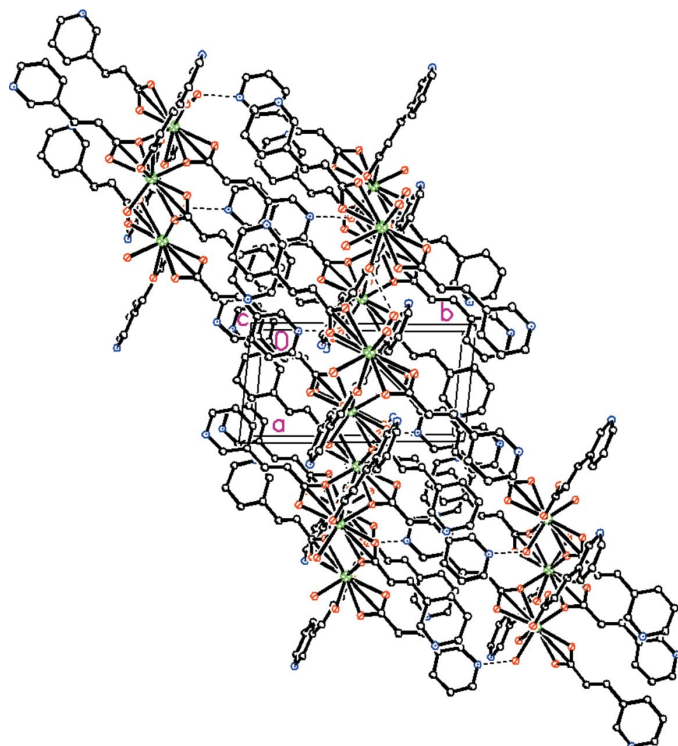


Figure 2

A packing diagram, viewed along the *c* axis. All H atoms have been omitted for clarity. Dashed lines indicate hydrogen bonds.

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